

Preparation and Cyclization of 4-Chloro-5,5-dimethyl-3-formyl-1,2-oxathiolene 2,2-dioxide

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ABSTRACT: Chloroformylation of 5,5-dimethyl-1,2-oxathiolan-4-one 2,2-dioxide **4** with Vilsmeier reagent (DMF/POCl₃) led to the formation of cyclic β -chlorovinylaldehyde (4-chloro-5,5-dimethyl-3-formyl-1,2-oxathiolene 2,2-dioxide **5**). Compound **5** reacted with formamidine, *o*-aminophenol, 1,2-phenylenediamine, aminopyrazole, and aminotetrazole to give the corresponding heterocyclic compounds. © 2005 Wiley Periodicals, Inc. *Heteroatom Chem* 16:200–204, 2005; Published online in Wiley InterScience (www.interscience.wiley.com). DOI 10.1002/hc.20094

INTRODUCTION

The Vilsmeier reagent, HCONR¹R²/POCl₃, has found extensive application in the synthesis of aldehyde derivatives and formamidines [1]. The reagent is an equilibrium mixture of two iminium salts, the more reactive being the β -phosphoryliminium chloride (A) rather than the β -chloroiminium phosphate (B) [2] (Fig. 1).

We recently reported that application of the Vilsmeier's reaction to acetylphosphonate led stereospecific formation of (*Z*)- β -phosphonyl- β -chlorovinylaldehyde [3]. We investigated this synthesis more deeply and widened it to 5,5-dimethyl-1,2-oxathiolan-4-one 2,2-dioxide **4**. Chloroformylation

of 5,5-dimethyl-1,2-oxathiolan-4-one 2,2-dioxide **4** with Vilsmeier reagent (DMF/POCl₃) afforded successfully cyclic β -chlorovinylaldehyde **5** (4-chloro-5,5-dimethyl-3-formyl-1,2-oxathiolene 2,2-dioxide). **5** is a very useful intermediate for the synthesis of heterocyclic compounds.

RESULTS AND DISCUSSION

The action of DMF/POCl₃ on 5,5-dimethyl-1,2-oxathiolan-4-one 2,2-dioxide **4** at 30°C led to the sole product 4-chloro-5,5-dimethyl-3-formyl-1,2-oxathiolene 2,2-dioxide **5** in 64.5% yield (Scheme 1). The reaction of **5** with formamidine, aminopyrazole, aminotetrazole, *o*-aminophenol, 1,2-phenylenediamine, and 5-phenyl-4-amino-3-mercapto-(4*H*)-1,2,4-triazoles gave the corresponding heterocyclic compounds **6–10** in the presence of K₂CO₃ (Scheme 2). These heterocyclic compounds were all characterized by their ¹H, ¹³C NMR, and elemental analysis.

Of all the new compounds, compounds **9a**, **9b**, and **10** have a seven-membered ring, which is commonly considered unfavorable because of the high strain in seven-membered ring. X-ray diffraction analysis of a single crystal of **10** further established the structure [4]. As depicted in Fig. 2, there undoubtedly exists a seven-membered ring C₁₁-C₁₀-C₉-S₁-C₈-N₃-N₄. Selective bond distances and angles are listed under Fig. 2.

On treatment of 4-chloro-5,5-dimethyl-3-formyl-1,2-oxathiolene 2,2-dioxide **5** with 1,2-phenylenediamine and aminopyrazole in the presence of K₂CO₃,

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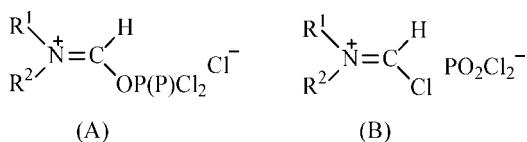


FIGURE 1 The two forms of Vilsmeier reagent.

which gave the expected compounds **7a** and **9b**, respectively, we also obtained small amounts of the imines **11** and **12** that resulted by the reaction of the formyl group of **5** with the amino group of aminopyrazole or 1,2-phenylenediamine (Scheme 3).

As shown in Scheme 3, a plausible pathway for this reaction would consist of two steps: in the first step, 1,2-addition of the amino groups of aminopyrazole or 1,2-phenylenediamine to the formyl group of **5** gives imines **11** and **12**; in the second step, intramolecular nucleophilic substitution of the nitrogen atom in **11** or **12** with chlorine atom on C₄ in them forms the final compounds **7a** and **9b**.

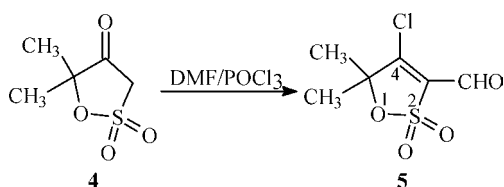
The reaction of 4-chloro-5,5-dimethyl-3-formyl-1,2-oxathiolene 2,2-dioxide **5** with hydrazine, hydroxylamine in the presence of K₂CO₃ can only give the intermediates hydrazones **13** and oxime **14** (Scheme 4), which cannot cyclize further to give the corresponding pyrazoles or isoxazole in the presence of NaH, K₂CO₃, or DMAP.

CONCLUSIONS

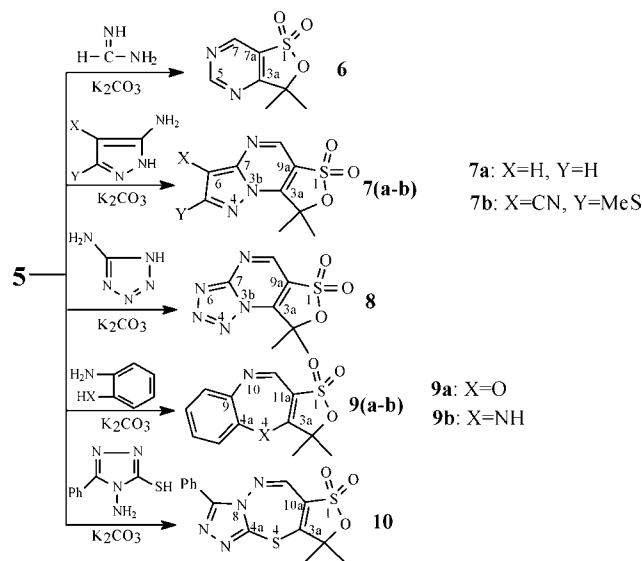
In conclusion, chloroformylation of 5,5-dimethyl-1,2-oxathiole-4-one 2,2-dioxide **4** with Vilsmeier reagent (DMF/POCl₃) led to 3-carbaldehyde-4-chloro-5,5-dimethyl-1,2-oxa-thiolene 2,2-dioxide **5**. Compound **5** is a very useful intermediate in organic synthesis. It reacted with formamidine, *o*-aminophenol, 1,2-phenylenediamine, aminopyrazole, and aminotetrazole to give the corresponding heterocyclic compounds in high yields.

EXPERIMENTAL

Melting points were determined with a Thomas-Hoover melting points apparatus. ¹H NMR spectra



SCHEME 1



SCHEME 2

were determined with a Bruker AC-P30 in CDCl₃ (CD₃)₂CO or DMSO solution. Chemical shifts were reported in ppm (δ) downfield from Me₄Si. Elemental analyses were performed on a Yanaco CHN Corder MT-3 apparatus.

Synthesis of Compound 4

To a mixture of acetone cyanohydrin **1** (4.0 mL, 43.8 mmol) in dry CH₂Cl₂ (25 mL) and Et₃N (12 mL, 87.5 mmol) was added dropwise methane sulfonyl chloride (4.8 mL, 62.1 mmol) whilst maintaining the temperature between 0 and 5°C. The reaction was allowed to reach room temperature and was stirred

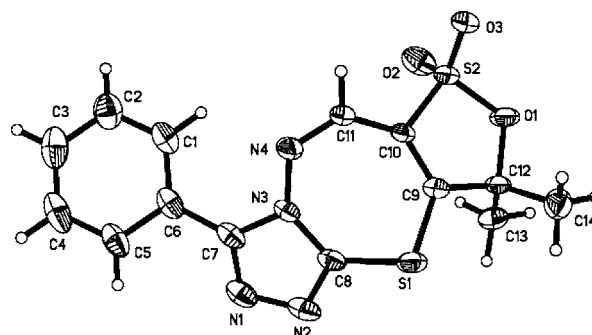
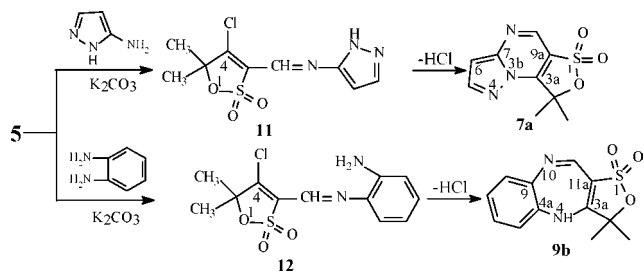


FIGURE 2 X-ray structure of compound **10**. The hydrogen atoms are omitted for clarity. Selective bond distance (Å) and angles (°): S(1)–C(8) = 1.723(4), S(1)–C(9) = 1.748(3), C(9)–C(10) = 1.338(4), C(10)–C(11) = 1.448(3), C(11)–N(4) = 1.275(4), N(4)–N(3) = 1.389(4). C(10)–C(11)–N(4) = 131.5(3), C(11)–N(4)–N(3) = 118.2(3), C(8)–S(1)–C(9) = 98.58, S(1)–C(9)–C(10) = 123.9(3).



SCHEME 3

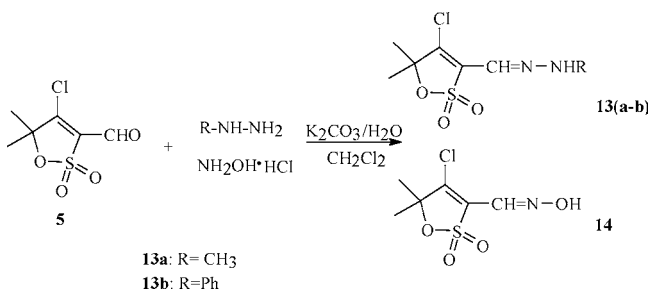
for 2 h. The mixture was filtered over Celite to remove triethylamine hydrochloride, and the filtrate was evaporated. Column chromatography (2:1 mixture of petroleum ether and AcOEt as the eluent) of the residue yielded compound **2** (5.30 g, 74%) as a colorless oil.

Compound **2** (0.43 g, 2.65 mmol) in dry THF (12 mL) was stirred with 60% NaH dispersion in oil (0.13 g, 3.18 mmol, 1.2 equiv) for 3 h at r.t. Water was added slowly to destroy the excess NaH, and the mixture was extracted with AcOEt (3 × 30 mL). The dried organic layer (MgSO₄) was evaporated to yield a white solid **3** that was recrystallized from CH₂Cl₂:acetone = 2:1 as colorless plates (0.305 g, 71%); mp 175–176°C (lit [5] mp 175–176°C).

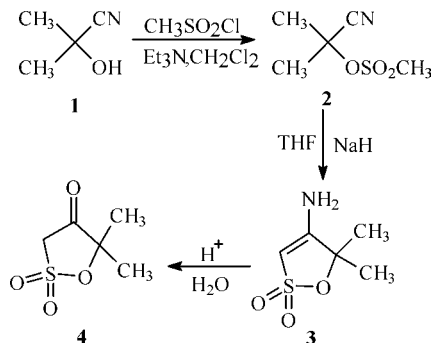
To a solution of **3** (0.305 g, 1.87 mmol) in EtOH (10 mL) was added concentrated HCl (1 mL) and then the mixture was stirred for 5 min. The mixture was filtered over Celite, and the filtrate was evaporated to leave a solid residue that, upon recrystallization from hexane:toluene (2:1), gave **4** (0.21 g, 66%) as colorless plates; mp 58–59°C (lit [4] mp 59°C) (see Scheme 5).

The Preparation of 4-Chloro-5,5-dimethyl-3-formyl-1,2-oxathiolene 2,2-dioxide (**5**)

To a mixture of POCl₃ (0.12 mol) in CH₂Cl₂ (15 mL) was added dropwise a solution of DMF (0.12 mol) in CH₂Cl₂ (5 mL) at 0°C under N₂ atmosphere. The



SCHEME 4



SCHEME 5

mixture was stirred at room temperature for 0.5 h. Then 5,5-dimethyl-1,2-oxathiolan-4-one 2,2-dioxide **4** (3.28 g, 0.02 mol) was added dropwise at 0°C. The reaction mixture was stirred for 30 h at room temperature. The mixture was poured slowly through a condenser to a bottle which contained 150 g crushed ice, and stirred for 3 h more. The aqueous layer was then extracted with CH₂Cl₂ (3 × 30 mL). The combined organic layers were washed with 50 mL brine, dried over Na₂SO₄, filtered and concentrated in vacuo. The crude yellow product was purified by recrystallization from AcOEt : petroleum ether (1:2) to give **5** as white crystals (2.73 g, 64.5%), mp 175–176°C; ¹H NMR (CDCl₃, 300 MHz): δ 9.88 (s, 1H, –CHO), 1.75 [s, 6H, (CH₃)₂]; ¹³C NMR (300 MHz, CDCl₃): δ = 178.22 (–CHO), 159.64 (C₅), 121.55 (C₄), 90.12 [(CH₃)₂–C–O], 26.78 (CH₃–); Anal. Calcd for C₆H₇SO₄Cl: C, 34.21; H, 3.35 Found: C, 34.15; H, 3.53.

General Procedure for the Cyclization of Compound **5**

To a mixture of compounds **5** (1 mmol) in CH₂Cl₂ (5 mL) was added dropwise a solution of formamidine, aminopyrazole [6,7], aminotetrazole [8], *o*-aminophenol, 1,2-phenylenediamine, or 5-phenyl-4-amino-3-mercapto-(4*H*)-1,2,4-triazole [9], methylhydrazine [10,11], phenylhydrazine, and hydroxylamine hydrochloride (1 mmol) in CH₂Cl₂ or water at 10°C, then K₂CO₃ (1 mmol) in water (2 mL) was added dropwise. The reaction mixture was kept at 30°C for 2–3 h. The aqueous layer was extracted with CH₂Cl₂ (3 × 10 mL). The combined organic layers were washed with 20 mL saturated brine, dried over Na₂SO₄, filtered, and concentrated. The residue was separated by silica gel using AcOEt : petroleum ether (3:1) as developing solvent to afford corresponding heterocyclic compounds **6–14**, which can be purified by recrystallizing from suitable solvents.

3,3-Dimethyl-2-oxa-1-thia-4,6-diaza-indan 1,1-dioxide (6). White crystals, yield 46.36%, mp 128–129°C; ¹H NMR (CDCl₃, 300 MHz): δ 9.45 (s, 1H, H₇), 9.29 (s, 1H, H₅), 1.85 [s, 6H, (CH₃)₂]; ¹³C NMR (CDCl₃): δ 170.31 (C_{3a}), 162.25 (C₅), 151.93 (C₇), 125.45 (C_{7a}), 93.09 (C₃), 26.21 (CH₃–); Anal. Calcd for C₇H₈N₂SO₃: C, 41.99; H, 4.03; N, 13.99. Found: C, 41.95; H, 4.08; N, 13.93.

3,3-Dimethyl-8H-2-oxa-1-thia-3b,4,8-triaza-indacene 1,1-dioxide (7a). Yellow crystals, yield 50.2%; mp 213–215°C; ¹H NMR (CDCl₃, 300 MHz): δ 9.21 (s, 1H, H₉), 8.41 (s, 1H, H₅), 6.86 (s, 1H, H₆), 1.94 [s, 6H, (CH₃)₂]; ¹³C NMR (CDCl₃): δ 160.35 (C_{3a}), 149.68 (C₉), 149.3 (C₇), 132.57 (C₅), 114.54 (C_{9a}), 98.19 (C₆), 93.44 (C₃), 26.63 (CH₃–); Anal. Calcd for C₉H₉N₃SO₃: C, 39.21; H, 3.66; N, 15.24. Found: C, 39.19; H, 3.76; N, 15.37.

6-Cyano-3,3-dimethyl-5-methylsulfanyl-1,3-dihydro-2-oxo-1-thia-3b,4,8-triaza-indacene 1,1-dioxide (7b). White crystals, yield 61.29%, mp 237–239°C; ¹H NMR (CDCl₃, 300 MHz): δ 7.05 (s, 1H, H₉), 2.75 (s, 3H, –SCH₃), 1.86 [s, 6H, (CH₃)₂]; ¹³C NMR (DMSO): δ 164.06 (C_{3a}), 162.09 (C₉), 152.72 (C₇), 136.01 (C₅), 122.36 (C_{9a}), 115.46 (–C≡N), 111.79 (C₆), 92.59 (C₃), 25.95 (–2CH₃), 13.18 (CH₃S–); Anal. Calcd for C₁₁H₁₀N₄S₂O₃: C, 42.57; H, 3.25; N, 18.05. Found: C, 42.40; H, 3.39; N, 17.98.

3,3-Dimethyl-8H-2-oxa-1-thia-3b,4,5,6,8-pentaza-indacene 1,1-dioxide (8). White crystals, yield 60.50%, mp 160–162°C; ¹H NMR (CDCl₃, 300 MHz): δ 8.91 (s, 1H, H₉), 1.82 [s, 6H, (CH₃)₂]; ¹³C NMR (300 MHz, CDCl₃): δ 173.75 (C_{3a}), 166.30 (C₇), 154.33 (C₉), 121.16 (C_{9a}), 92.40 [(CH₃)₂–C–O], 26.11 (–2CH₃); Anal. Calcd for C₇H₇N₅SO₃: C, 34.85; H, 2.93; N, 29.03. Found: C, 34.77; H, 2.88; N, 28.93.

3,3-Dimethyl-3H-2,4-dioxo-1-thia-10-aza-benzoazulene 1,1-dioxide (9a). White crystals, yield 45.36%, mp 171–172°C; ¹H NMR (CDCl₃, 300 MHz): δ 7.23–6.98 (m, 4H, Ph), 5.54 (s, 1H, H₁₁), 1.76 [s, 6H, (CH₃)₂]; ¹³C NMR [(CD₃)₂CO]: 157.07 (C_{3a}), 149.64 (C₁₁), 128.03, 127.26, 124.65, 120.70, 117.04 (Ph), 115.57 (C_{10a}), 88.25 [(CH₃)₂–C–O], 26.12 (CH₃–); Anal. Calcd for C₁₂H₁₁SO₄: C, 54.33; H, 4.18; N, 5.28. Found: C, 54.47; H, 4.18; N, 5.14.

3,3-Dimethyl-3,4-dihydro-2-oxa-1-thia-4,10-diaza-benzoazulene 1,1-dioxide (9b). Purple crystals, yield 56.82%; mp 172–173°C; ¹H NMR (CDCl₃, 300 MHz): δ 6.71–6.78 (m, 2H, Ph), 6.54–6.58 (d, *J* = 6.58 Hz, 1H, H₈), 6.37 (s, 1H, H₁₁), 5.98–6.02 (d, *J* = 6.58 Hz, 1H, H₅), 1.59 [s, 6H, (CH₃)₂]; ¹³C NMR

[(CD₃)₂CO]: δ 171.94 (C_{3a}), 154.12 (C₁₁), 133.34, 130.03, 129.23, 124.92, 123.19, 121.19 (Ph), 117.95 (C_{11a}), 90.90 [(CH₃)₂–C–O], 25.68 (CH₃–) Anal. Calcd for C₁₂H₁₂N₂SO₃: C, 54.49; H, 4.57; N, 10.60. Found: C, 54.26; H, 4.62; N, 10.79.

3,3-Dimethyl-7-phenyl-3H-2-oxa-1,4-dithia-5,6,8,9-tetraaza-cyclopenta-azulene 1,1-dioxide (10). Yellow crystals, yield 72.5%; mp 215–217°C (decompose); ¹H NMR (CDCl₃, 300 MHz) δ 7.86–7.92 (m, 3H, Ph), 7.51–7.57 (m, 3H, Ph, and CH=N) 1.83 [s, 6H, (CH₃)₂]; ¹³C NMR (CDCl₃): δ 161.19 (C₁₀), 157.15 (C_{3a}), 142.59 (C₈), 139.99 (C₅), 131.24, 129.51, 128.65, 126.92 (Ph), 124.74 (C_{10a}), 90.47 [(CH₃)₂–C–O], 25.93 (–2CH₃); Anal. Calcd for C₁₄H₁₂N₄S₂O₃: C, 48.26; H, 3.47; N, 16.08. Found: C, 48.14; H, 3.50; N, 16.12.

(4-Chloro-5,5-dimethyl-2,2-dioxo-2,5-dihydro-[1,2]oxathiol-3-ylmethylene)-(2H-pyrazol-3-yl)-amine (11). White crystals, yield 15.8%; mp 217–219°C; ¹H NMR (CDCl₃, 300 MHz) δ 9.21 (s, 1H, NH=N=CH), 8.40 (s, 1H, –CH=N), 6.86 (s, 1H, C=CH), 1.94 [s, 6H, (CH₃)₂]; Anal. Calcd for C₉H₁₀N₃SO₃: C, 45.18; H, 3.79; N, 17.56. Found: C, 45.22; H, 3.86; N, 17.33.

N-(4-Chloro-5,5-dimethyl-2,2-dioxo-2,5-dihydro-[1,2]oxathiol-3-ylmethylene)-benzene-1,2-diamine (12). Yellow crystals, yield 26.67%, mp 138–139°C; ¹H NMR (CDCl₃, 300 MHz) δ 8.35 (s, 1H, H=C=N), 7.08–7.13 (m, 2H, Ph), 6.62–6.71 (d, *J* = 7.68 Hz, 1H, Ph), 6.62–6.67 (d, *J* = 7.68 Hz, 1H, Ph), 2.93 (s, NH₂), 1.73 [s, 6H, (CH₃)₂]; Anal. Calcd for C₁₂H₁₃N₂SO₃Cl: C, 47.74; H, 4.48; N, 9.42. Found: C, 47.92; H, 4.36; N, 9.31.

N-(4-Chloro-5,5-dimethyl-2,2-dioxo-2,5-dihydro-[1,2]oxathiol-3-ylmethylene)-N'-methyl-hydrazine (13a). White crystals, yield 56.8%, mp 130–131°C; ¹H NMR (CDCl₃, 300 MHz): δ 6.94 (s, 1H, CH=N), 2.97 (s, CH₃N–) 1.67 [s, 6H, (CH₃)₂]; ¹³C NMR (CDCl₃): δ 154.7 (CH=NNHCH₃), 138.85 (C₄), 116.14 (C₃), 89.07 (C₅), 32.65 (NH–CH₃), 25.74 (2CH₃); Anal. Calcd for C₇H₁₁N₂SO₃Cl: C, 35.22; H, 4.66; N, 11.73. Found: C, 34.87; H, 4.98; N, 11.73.

N-(4-Chloro-5,5-dimethyl-2,2-dioxo-2,5-dihydro-[1,2]oxathiol-3-ylmethylene)-N'-phenyl-hydrazine (13b). Yellow crystals, yield 67.5%, mp 161–162°C; ¹H NMR (CDCl₃, 300 MHz): δ 8.21 (s, 1H, NH), 7.36 (s, CH=N), 7.30–6.90 (m, 5H, Ph), 1.69 [s, 6H, (CH₃)₂]; ¹³C NMR (CDCl₃): δ 142.48 (CH=NNH), 138.91 (C₄), 121.29 (C₃), 129.62, 129.39, 122.14, 113.64 (Ph), 89.30 (C₅), 25.70 (2CH₃); Anal. Calcd

for $C_{12}H_{13}N_2SO_3Cl$: C, 31.94; H, 3.57; N, 6.21. Found: C, 32.01; H, 3.66; N, 6.17.

4-Chloro-5,5-dimethyl-2,2-dioxo-2,5-dihydro-[1,2]oxathiole-3-carbaldehyde oxime (14). White crystals, yield 68.9%, mp 154–155°C; 1H NMR ($CDCl_3$, 300 MHz): δ 8.64 (s, 1H, $-OH$), 7.94 (s, $\underline{CH=N}$), 1.71 [s, 6H, $(\underline{CH}_3)_2$]; ^{13}C NMR ($CDCl_3$): δ 146.87 ($\underline{CH=NOH}$), 137.76 (C_4), 127.14 (C_3), 90.32 (C_5), 25.51 (2 CH_3). Anal. Calcd for $C_6H_8NSO_4Cl$: C, 47.92; H, 4.36; N, 9.31. Found: C, 48.08; H, 4.46; N, 9.86.

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